

# CLASS ACTIVITY

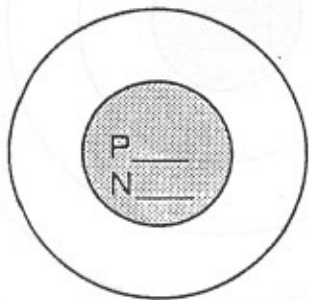
NAME \_\_\_\_\_ PER. \_\_\_\_\_ DATE: \_\_\_\_\_

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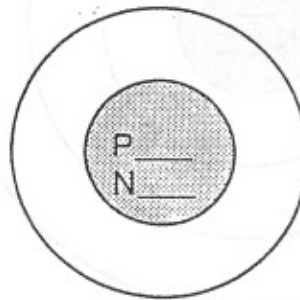
## STRUCTURE OF ATOMS

**NAME THE ATOM, THEN INDICATE THE NUMBER OF PROTONS, NEUTRONS, AND ELECTRONS.**

1.



2.

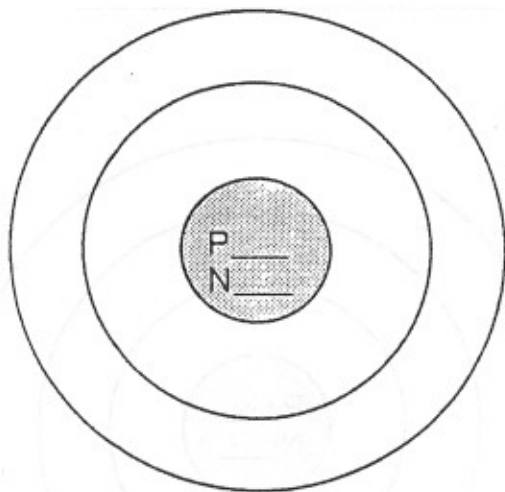


e = ●

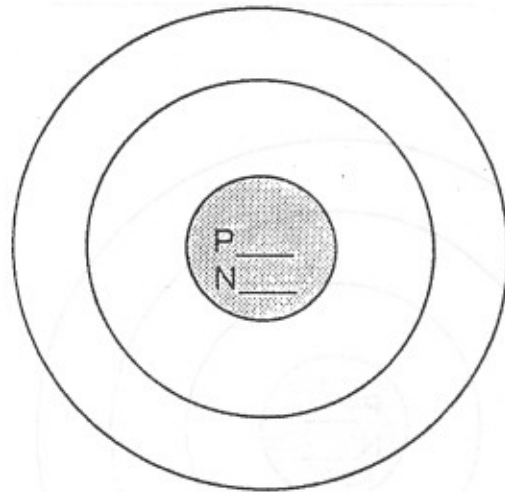
H is \_\_\_\_\_

He is \_\_\_\_\_

3.



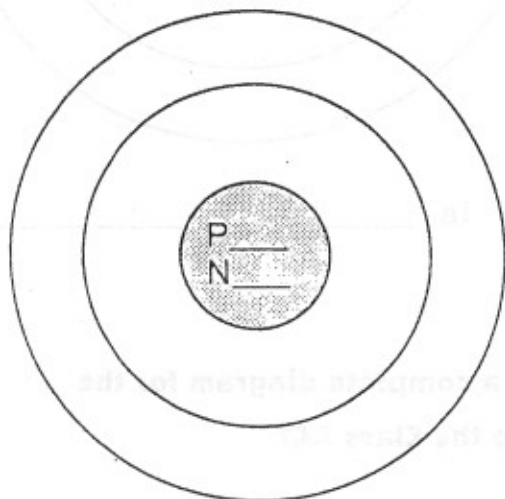
4.



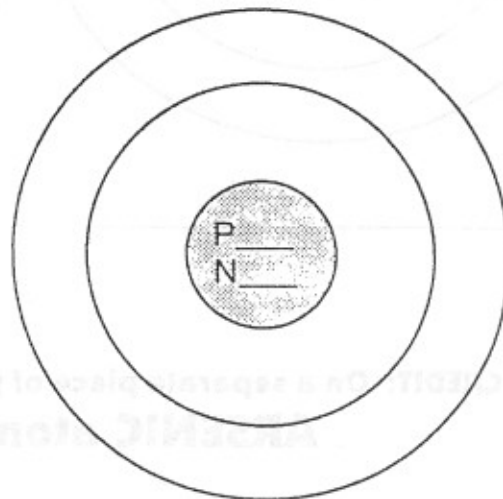
Li is \_\_\_\_\_

B is \_\_\_\_\_

5.



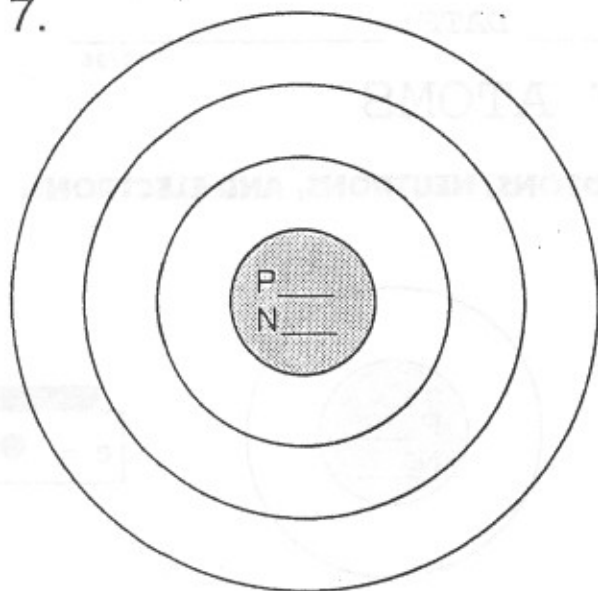
6.



N is \_\_\_\_\_

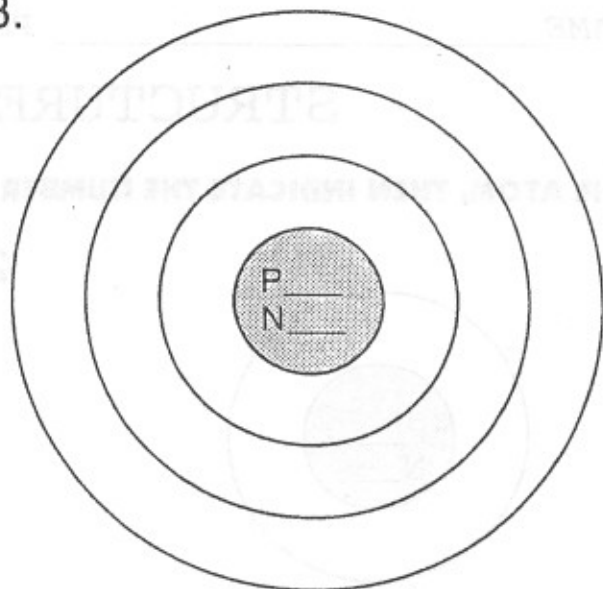
Ne is \_\_\_\_\_

7.



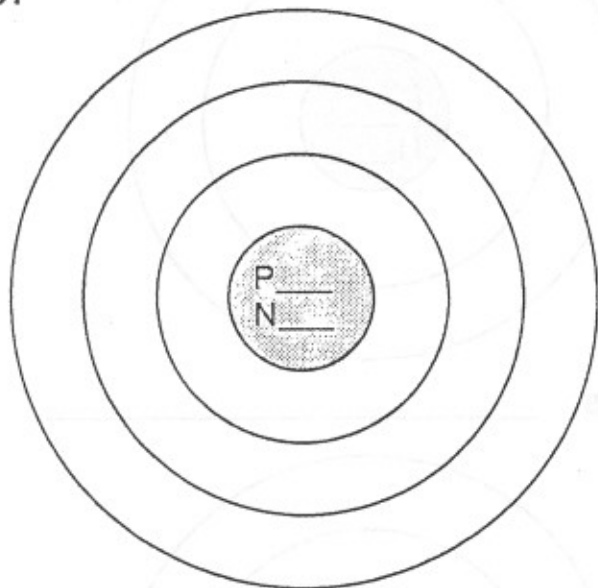
Na is \_\_\_\_\_

8.



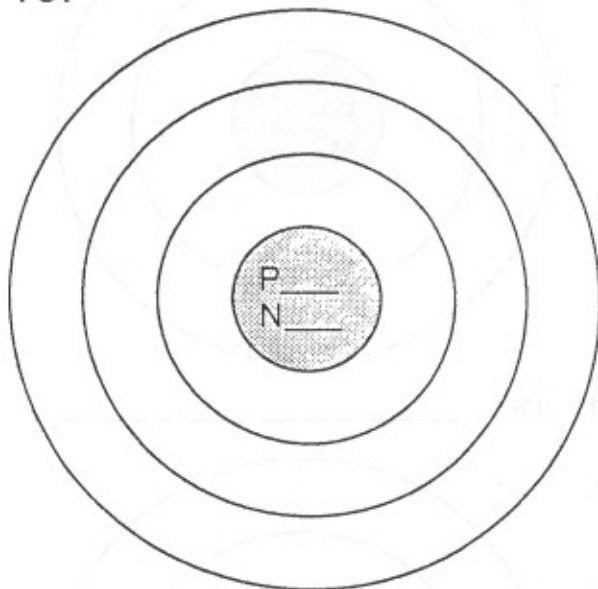
Al is \_\_\_\_\_

9.



P is \_\_\_\_\_

10.



S is \_\_\_\_\_

**EXTRA CREDIT:** On a separate piece of paper draw a complete diagram for the **ARSENIC** atom. Attach to the Class ACT.